

ERE MESSENGER

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Spring 2003

ERE Curriculum Changes: Engineering Design and Computational Methods

by
C. Mike Anderson, PE
ERE Professor and Chair

Effective Spring 2003, there are changes to two areas of the undergraduate ERE curriculum: engineering design and computational methods. The numbers of required courses and units remain unchanged, but some content and distribution changes are significant. The changes were approved too late to appear in the 2002-2003 printed catalog, but we expect them to be posted to the HSU On-Line Catalog at <www.humboldt.edu/~catalog> by mid February 2003.

Engineering Design

We replaced our old elective course ENGR 490 "Senior Design Project" with a new required course ENGR 492 "Capstone Design Project." The primary purpose of ENGR 492 is to demonstrate to students, faculty, potential employers and grad schools,

and the Accreditation Board for Engineering and Technology (ABET) that all ERE graduates have the ability to apply the engineering design process to solve complex, broad-based environmental resource problems. ENGR 492 focuses entirely on the engineering design process and its application to solve real ERE problems. Capstone (or culminating) is a key word here; ENGR 492 is the last in a series of course containing engineering analysis and design. Projects will involve multiple aspects of air-water-land-energy issues learned from previous required and elective courses. Broad project themes include: prevention and control of pollution, allocation and management of resources (especially scarce resources), and sustainable development.

Under the old scheme, students selected four senior design electives from an approved list. Under the new

scheme, students select three senior design electives from the approved list, plus all students take ENGR 492. In selecting their three elective senior design courses, students choose the topics they wish to cover. Students in ENGR 492 do not choose the topic, but rather show that they are competent in the broad areas of their degree title, Environmental Resources Engineering. This change will make it much easier for us to demonstrate to ABET that all of our students have received the required capstone design experience.

Finally, we added a new senior

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ENGR 492. Capstone Design Project. 3 units. Culminating ERE design experience based on knowledge gained from previous coursework. Application of the engineering design process to develop a system, process or management plan to solve a significant, open-ended ERE problem. To be taken final senior semester (within 16 units of graduation).

Alumni Profiles

Joaquin Wright (ERE 2002)

Engineer

Bryan A. Stirrat & Associates

Diamond Bar, CA 91765

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I often wondered what I would really be doing after school was over. Although my classes were interesting, it was hard to see how they would pertain to a "real" job. I ended up working for Bryan A. Stirrat and Associates in the landfill business, you know... Garbage. In this business our projects encompass almost all the "disciplines" within environmental engineering.

To help current ERE students see what they might expect in our field, in the real world, I have summarized below two of the projects I am currently working on.

Belmont Learning Complex in Los Angeles, California.

We are designing and installing a Methane extraction and removal system on 40 acres of residential land in a neighborhood directly north of downtown Los Angeles. A methane extraction system is to be installed under the new school at the site, Belmont Learning Complex.

The project was resurrected after it had been shut down in mid stream, around 1998. The first construction companies did not account for environmental concerns to a level appropriate, and as a consequence

the construction was halted. Four years later, with a new team of constructors and engineering firms, it has been restarted and construction is scheduled for April 2003. We have been hard at work developing a system that will insure no migrating gas from the old oil fields comes up into the school buildings, fields or play areas. It might sound simple, but it is quite a challenge to develop membrane and extraction systems that are going to be installed under and into existing and planned buildings. We have to account for all the underground utilities, keeping our system below them and allowing for our system to drain as well; it is an engineering design challenge. Along with the extraction system we also are putting in detection devices and blowers that can evacuate all the buildings of "contaminated" air instantly, if the need arises. During the design process we have other engineers and geologists drilling at the site and taking soil samples, water samples, and running tests to get information that will help us design the correct system for the site. I am one of four design engineers focusing on the membrane and gas removal system. In the end, our portion alone will be approximately \$4 million. The total project is estimated to cost at least \$80 million. Needless to say, if we need something to help us do our work, we get it.

Prima Deschutes Landfill in Irvine, California.

Prima has been the most interesting of my projects. During the permitting involved with the local government to expand the site (something that happens every year or so)

all the various government agencies have their input. As a result of the last go around, we mitigated many riparian areas, rebuilt a stream and its relative ecosystem around the site, and we designed and constructed a new area of expansion, which we call "cells". In order to move and rebuild a stream in the hills, every type of biologist, environmental engineer, soils engineer, and politician had a part to play.

In the end I got to design the stream and its basic structure (under a much more experienced engineer) as we rerouted the stream down the mountain away from the future cell. Due to natural constraints (the mountain) we had to build into it waterfalls and energy dissipation structures as well as riparian areas and marsh areas. All of this was accomplished under the eye of nearly every regulatory agency that ever existed!!! It was a neat project to be part of.

To end my time on the soap box, here is one piece of advice... Listen up when your instructors talk about the projects they have worked on. This is the best information you can glean from school. It is where the true gems of information are that can shed some light as to what we actually do as environmental engineers. Personally, I find it very fulfilling. I am very proud to have survived the meat grinder I call ERE at HSU. It was and is no cakewalk, as we all know, but it has served me well.

If you plan to head for southern California when you graduate, please send me your resume. I will do what I can to help you find opportunities. It is a small world and the more HSU ERE graduates that work in it the better we will all be. **ERESA**

S T A F F

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To the optimist, the glass is half full.

To the pessimist, the glass is half empty.

To the engineer, the glass is twice as big as it needs to be.

INTERNSHIPS FOR ERE STUDENTS

by
Barbara Peters
Career Counselor
HSU Career Center

What is an internship and why should you do one? An internship is defined as an "out of classroom" experience related to the student's major or career goals. The position can be a volunteer position, offer a stipend (a fixed amount of pay), or provide an hourly wage; positions can be offered part-time during the school year or full-time over the summer. Some employers offer "summer jobs" that can even fall under the "internship" definition because they relate to the student's major.

The importance of completing an internship in your field cannot be stated too strongly. In my 26+ years of working with college students, I am constantly reminded of the value a student gains from working in his/her field over the summer or during the school year. In fact, in the "Survey of Graduates" that the HSU Career Center conducts every year, "work/volunteer/internship experience" is consistently listed as the most important factor that graduates mention in assisting them to obtain their first job after graduation. An internship can help the student focus their educational and career goals, apply what they are learning in the classroom to actual engineering projects, gain experience, and make contacts which are invaluable when it comes to job hunting after graduation.

The best time to start researching and applying for summer internships is early in the Spring semester, although some internships and summer jobs will not be advertised until May. Some samples of internships that have been available for ERE stu-

dents in the past include the following:

- Indian Health Service (PHS) – Student Engineer (COSTEP) to design and construct water, wastewater, and solid waste facilities for Indian communities throughout the country. Deadline is December 31, but applications accepted year-round.
- Environmental Protection Agency – Waste Management Intern. Offered through the Environmental Careers Org. (www.eco.org).
- CalTrans – Engineering Student Assistant. Various locations throughout northern California. Deadline varies – usually in March
- Calif Dept of Water Resources – Flood Mgt Student Assistant
- Idaho Department of Environmental Quality – Environmental Trainee
- City of Eureka – Engineering Aide
- Central Contra Costa Sanitary District – Engineering Intern (6 month position)
- County of Humboldt – Engr Aide. Deadline usually in March
- John Carollo Engineers (Walnut Creek) – Summer Engr Intern
- Foresight Energy Company (Larkspur, CA) – Solar/Efficiency Intern
- Golder Associates (Roseville, CA) – Engineering Student Intern
- Pacific Gas & Electric – Environmental Engineering Intern
- Pacific Soils Engineering – Summer Engineering Intern
- URS Corporation – Design Engineering Intern
- SHN Engineering – Environmental Engineering Intern

Current internships advertised through the HSU Career Center can

be accessed through the Career Center webpage (www.humboldt.edu/~career). After you register through MonsterTRAK, go to the "Jobs and Internships" section of MonsterTRAK. In the "keyword" block, type "Humintern SNRE/" – this will give you the current "Science, Natural Resources and Engineering" internships being advertised:

- Sonoma County Water Agency – NR Program Assistant
- Biogeochemical Research Initiative for Education (BRIE) – Undergraduate Fellowship. Deadline is February 28, 2003.
- Consumer Energy Council of America – Public Policy Intern. Internships available year-round.

Other opportunities currently available for ERE students can be found in binders in the HSU Career Center or on-line with the National Science Foundation (Research Experience for Undergraduate – REU positions) at www.nsf.gov

- Natural Resources Conservation Service – SCEP position in Engineering in California. Deadline extended to January 21, 2003.
- U.S. Forest Service – SCEP positions in Civil Engineering throughout U.S. – HSU deadline for interviewing on campus: January 31, 2003; deadline for applying directly to USFS regions. Deadline – February 28, 2003.
- Environmental Protection Agency – Student Temporary Employment Program (STEP) positions in Engineering in North Carolina. Deadline – March 28, 2003. (see www.epa.gov/jobs).
- Department of the Navy – Engr Aid/Tech (STEP) in Pacific Geographic area. Deadline: Open. (See www.usajobs.opm.gov).
- Colorado State University – REU program in Water Research. \$2400 stipend. Deadline – March 28, 2003.
- University of Idaho – REU program in Environmental Research for under-represented students. \$3528 stipend. Deadline – March 15, 2003.
- University of Notre Dame – REU

position in Water Resources in Developing Countries. Deadline - February 17, 2003.

- Washington University in St. Louis - REU position in Environmental Engineering Science Program. \$4000 stipend. Deadline - March 15, 2003.
- Rowan College of Engineering (New Jersey) - REU position in Pollution Prevention. \$1000 monthly stipend. Deadline - March 14, 2003.
- Clarkston University (New York) - REU position in Environmental Sciences and Engineering. \$3500 stipend. Deadline - March 1, 2003.
- North Carolina State University - REU position in Green Processing. Deadline - February 28, 2003.

As the semester continues, more internships and other summer opportunities will be advertised through MonsterTRAK, in the ERE Department, and on-line. The Career Center usually sends an e-mail advertisement to Mike Anderson and/or Beth Eschenbach in the ERE Department and they will forward these e-mails to students on their listservs. Students can also pursue setting up their own internships or summer opportunities with private engineering firms. Several resources are available to help students identify environmental engineering firms. The Career Center has a copy of the Selection Guide, published by the American Academy of Environmental Engineers. The American Consulting Engineers Council has a website (<http://206.135.109.15>) where students can search for engineering firms by topical area. And, students can also locate consulting firms in specific geographic areas by using the "Yellow Pages" of the telephone directory.

When applying for internships and summer jobs, you will need to fill out an employer's application, complete a federal resume (kept in a database at www.usajobs.opm.gov), or submit a resume and cover letter. Students can pick up a Resume Guide in the HSU Career Center (NHW 130). Barbara Peters, Career Counselor for the ERE major, is available by ap-

pointment to check over applications, resumes, and cover letters. Just stop by the HSU Career Center or call 826-3341 to schedule an appointment.

To summarize, here are some basic tips to remember when applying for internships and summer jobs:

- Start early in Spring semester and continue looking throughout the Spring.
- Prepare your resume, cover letter, and federal resume in advance.
- Monitor MonsterTRAK (www.humboldt.edu/~career) weekly for any newly advertised opportunities.
- Check the federal government website (www.usajobs.opm.gov), Student Jobs section, weekly for any new Engineering Aid/Tech positions.
- Check your e-mail regularly for any internships sent to you by the Career Center or faculty in your department.
- Watch the bulletin boards or doors of faculty offices.
- Do your own searches for summer internships on-line.
- Contact consulting engineering firms to set up your own internships.

Good luck and happy hunting! **ERESA**

ERE Curriculum Changes

• continued from page 1

design elective, ENGR 498 "Directed Design Project," to provide students the opportunity to earn design credit for independent projects under the direction of an ERE faculty member.

Computational Methods

We deleted three required 3-unit courses and replaced them with three new required 3-unit courses. The old courses are: MATH 225 "Applied Linear Algebra with Ordinary Differential Equations" and ENGR 221, 321 "Computational Methods I and II." The new courses are ENGR 225, 325, 326 "Computational Methods I, II, III." All three new courses have two 1-hour lectures and one 3-hour lab

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The Moment

by
Thavisak Syphanthong
ERE Senior

The Moment we met is like
The Moment of Inertia
When the two of us are standing
On this earth here
And our velocities are crossing the
control volume
Of love and despair
In hopes of crossing the northern
and southern
Hemisphere
Which is the sum of the torque
From London to New York
That would return vectors
Of \mathbf{r} and \mathbf{v}
That are the symbolizations of you
and me
And \mathbf{w} is the omega
Of a rotary water sprinkler
That sprinkles tears of rain
When doctors operate the brain
To dig up emotions
That would dry up oceans
And there are operating systems
That can calculate
The inflow of steam engines
And pistons
But they cannot compare
With the steam rising
When we are kissing
And they can program a program
To find pressure heads
Of wells that are missing
But they cannot derive something
as essential
As the Art of Kissing
And our Love is like
The flow of groundwater
That is u-surfaced from the earth
first
And purified through
A drinking water system
That is aggravated, coagulated
And agitated until the mixture is
complete
And we meet together as one
In the Moment that we met.

REMOVAL OF SELENIUM THROUGH REDUCTION IN ANAEROBIC SOILS: A COLUMN STUDY

by

Mark Leisenring (ERE 2001) and Jim Howell (ERE 1994)
GeoSyntec Consultants, Inc.
Portland, OR

Abstract: A static column study was conducted for three weeks in September 2002 to identify an effective soil medium for reducing selenium in the surface waters of San Diego Creek in Orange County, California. Six media mixtures were examined, each consisting of one or more of the following: native soil, silica sand, chopped cattails, municipal green waste, coconut fiber (coir), and sodium acetate. Wetland soil from the San Joaquin Marsh was used as the inoculum for each of the media, as it was believed that these anaerobic soils contained selenium-reducing bacteria. The results of the study indicate that the medium consisting of native soil, silica sand and chopped cattails was effective at removing soluble selenium from the source water and had other desirable characteristics (i.e. inexpensive, easy to obtain, no unwanted chemicals, etc.). This media mixture was chosen for further study in a mesocosm-scale experiment.

Background

Selenium concentrations in the base flows of San Diego Creek are frequently observed at levels that exceed the California Toxics Rule (CTR) chronic toxicity criteria for the protection of aquatic life (40 CFR §131.38). The current source of selenium in the watershed (Hibbs and Lee, 2000) is the historically reduced soils of the Swamp of the Frogs, which have been exposed to a fluctuating groundwater table ever since the swamp was drained for agriculture in the early 1900's. Exposure to oxygenated groundwater slowly releases soluble selenium from the historic soils of the swamp into the groundwater that eventually enters the creek.

The Irvine Ranch Water District (IRWD) has initiated a pilot study to examine the treatability of selenium-laden waters using a subsurface flow constructed wetland design that would function similarly to the historic Swamp of the Frogs by immobilizing selenium in reduced sediments. The pilot study was divided into two parts. The first is a laboratory-scale static column test, and the second is a field-scale dynamic mesocosm test. Limited results of the laboratory col-

umn test are presented in this article.

Objectives and Approach

The purpose of the column study was to identify an effective and efficient selenium-reducing medium for further study at the flow-through mesocosm level. Six soil mixtures, approximately proportioned by wet volume, were chosen for analysis:

- Medium 1: 90% native soil, 10% wetland soil;
- Medium 2: 60% native, 10% wetland soil, and 30% sand;
- Medium 3: 50% native soil, 30% silica sand, 10% wetland soil, 10% chopped cattails;
- Medium 4: 50% native soil, 30% silica sand, 10% wetland soil, and 10% green waste;
- Medium 5: 50% native soil, 30% silica sand, 10% wetland soil, and 10% coir; and
- Medium 6: 90% silica sand, 10% wetland soil, and 5 mmol sodium acetate.

Research on the biogeochemical behavior of selenium in anaerobic soils revealed that selenium reduction gen-

erally follows the sequential reduction (i.e. increasing oxidation-reduction potential (ORP) of dissolved oxygen, nitrate, manganese (IV), and iron (III)). Sulfate reduction generally follows selenate and selenite reduction (Masscheley et al., 1991). Consequently, field measurements of DO and ORP were taken daily to estimate the onset of reduction and determine when sampling for laboratory analysis would begin.

The columns were made out of 6-inch diameter polyvinyl chloride (PVC) cut to 5-foot lengths. Labcock valves installed in the bottom of the columns were used to collect samples. Six different detention times were examined for the six media mixtures under static conditions. A separate column was used for each media-detention time combination. Two laboratory samples (standard plus one replicate) were taken from two separate columns at the end of each scheduled detention time, resulting in 36 columns that were each sampled twice. The prepared media was placed into the columns, above a 4" bottom layer of #20 filtration sand. A mesh filter was used to prevent clogging of the port.

Water Quality Monitoring

The columns were filled with media and source water on August 26, 2002. On August 27 field monitoring began. Two field probes were placed into two of three inline cells for measurement of DO and ORP. Water temperature was also recorded. Due to very low flow rates observed in the first two media mixtures, it was decided to abandon these columns. On September 7, nearly two weeks after the beginning of the experiment, the columns containing the four remaining mediums were drained and refilled from the bottom with source water. The drain water was analyzed for the entire suite of laboratory parameters shown in Table 1. Official sampling for the analytical suite of parameters began on September 9 and followed every 2 days for a total of 5 sampling events.

Results

Dissolved organic carbon is a measure of the amount of electron donor available for microbial metabolism, which tends to catalyze the reduction of electron acceptors, such as oxygen, nitrate, selenium, manganese, iron, and sulfate. Figure 1 compares dissolved organic carbon to dissolved selenium for all of the col-

Parameters	
Dissolved organic carbon (mg/L)	Nitrate as N (mg/L)
Total organic carbon (mg/L)	Nitrite as N (mg/L)
Hydrogen ion (pH)	Dissolved oxygen (DO)
Iron (II) and Iron (III) (mg/L)	Dissolved selenium (mg/L)
Dissolved manganese (mg/L)	Total selenium (mg/L)
Total manganese (mg/L)	Sulfate (mg/L)

Oxidation-reduction potential (mV) columns on Sept 7. Notice the general inverse relationship between dissolved organic carbon and dissolved selenium, indicating that the lack of sufficient electron donors in Mediums 5 and 6 contributed to the relatively high dissolved selenium concentrations in the effluent.

Samples taken from Sept 9 through Sept 19 were analyzed for the entire suite of analytical parameters. The first samples analyzed revealed that Mediums 3 and 4 already had effluent selenium concentrations less than the laboratory reporting limit of 5 mg/l (these results were shown for discussion purposes only; any values reported below the reporting limit are not considered to be the true value). Dissolved selenium concentrations for each column of each me-

dia type are shown in Figure 2. Note that all of the columns had dissolved selenium concentrations below the reporting limit on the final sampling date (Sept 19). An interesting phenomenon observed in these data is the significant reduction in dissolved selenium in the second sample (replicate) taken from a column as compared to the first sample (standard). The reason for these "dips" in concentrations is undetermined. However, it is speculated that these reductions are due to the uneven distribution of electron donor and/or microbial populations caused by refilling the columns from the bottom with source water on Sept 7.

Notice that the samples taken from Medium 6 on Sept 11 and 13 had selenium concentrations greater

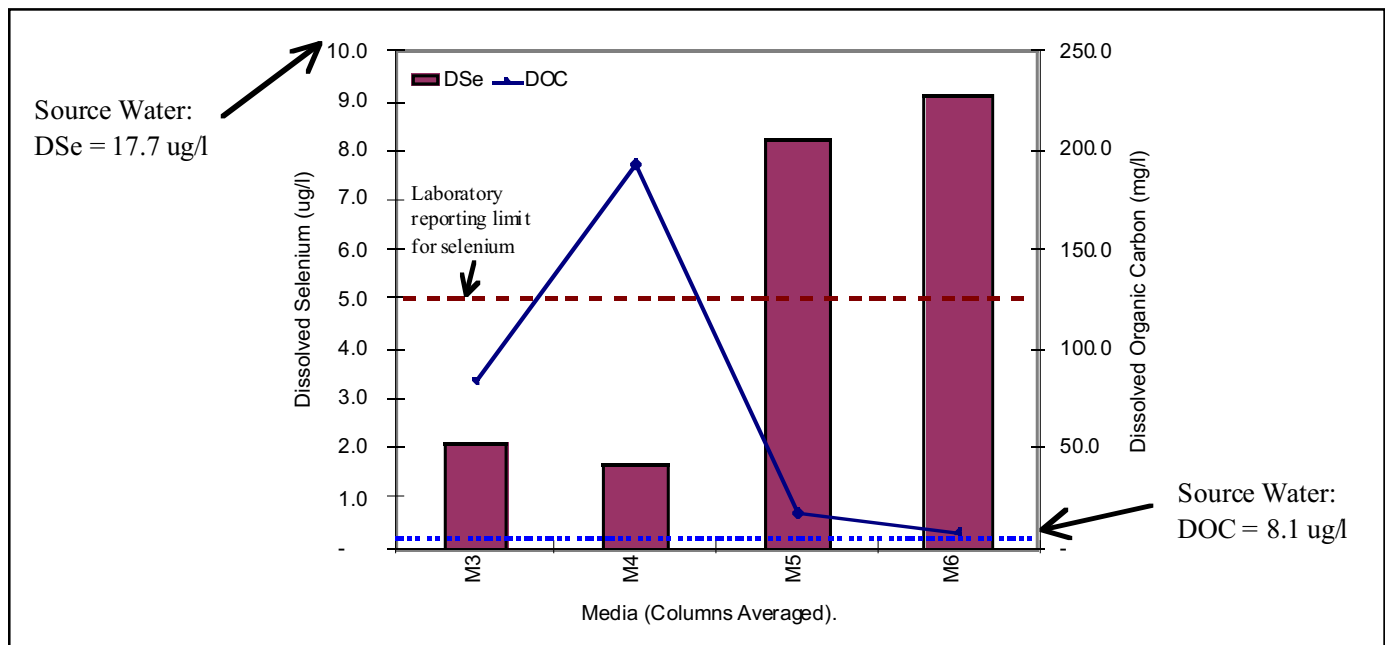


Figure 1. Dissolved Selenium vs Dissolved Organic Carbon (September 7, 2002)

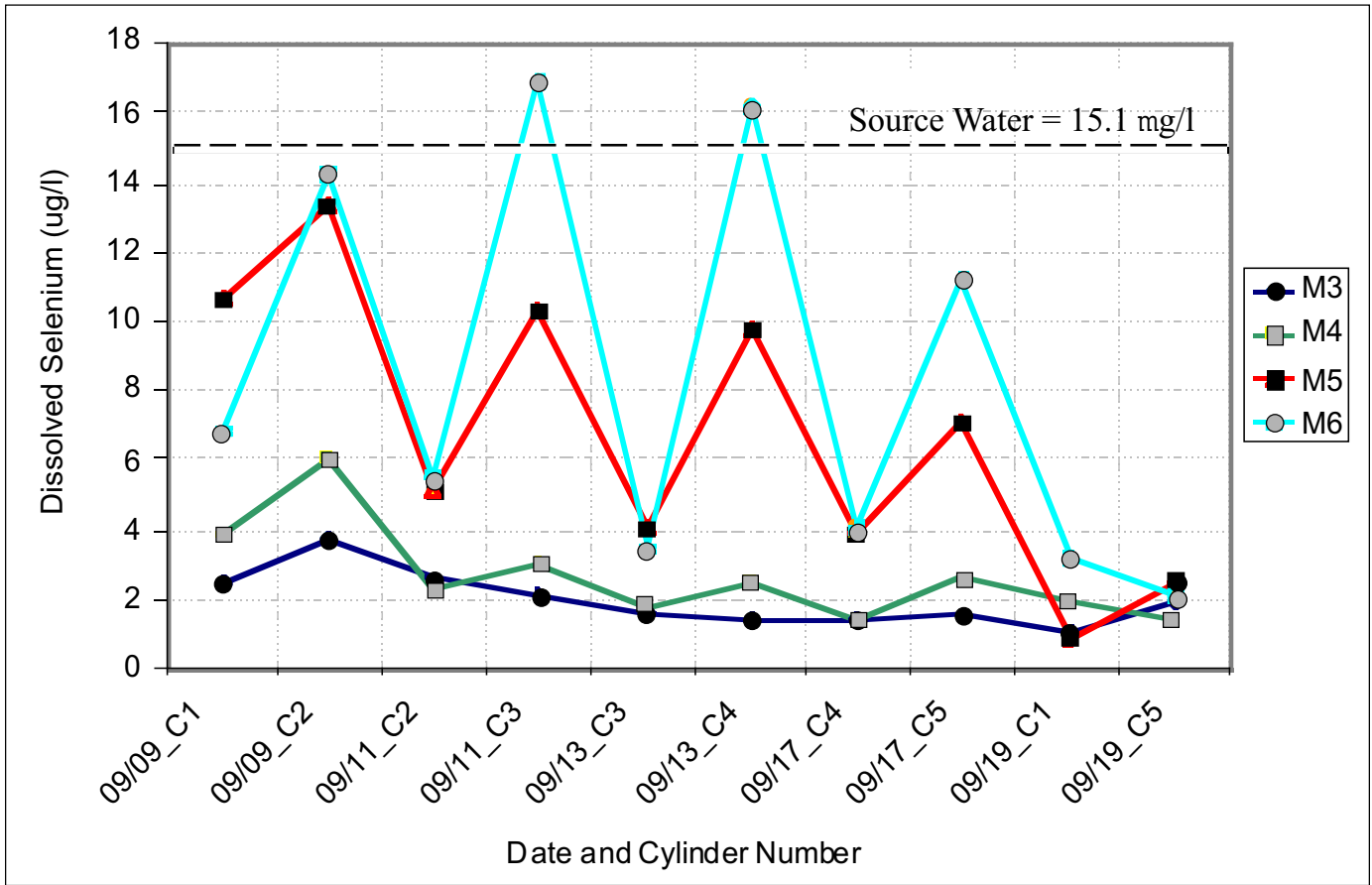


Figure 2. Dissolved Selenium Concentrations for Each Column for Each Media Type

than the source water. One possible explanation for these elevated concentrations is the influent water temporarily oxidized selenium that had been reduced and sequestered in the media prior to draining the columns on Sept 7.

To examine the selenium reduction recovery rate of each media type the difference between the average of the Sept 7 concentrations and the average of the Sept 9 concentrations relative to the difference between the existing concentrations (Sept 7) to the new source water concentration are shown in the Table 2. As can be seen from the table, Medium 3 was the least affected by the inundation of selenium-laden oxygenated waters.

Discussion and Conclusion

The results of the column study indicate that Medium 3 (50% native soil, 30% silica sand, 10% wetland soil, 10% chopped cattails) and Medium 4 (50% native soil, 30% silica sand, 10% wetland soil, and 10% green

waste) are the best selenium-reducing media mixtures of the four ultimately tested. The high dissolved organic carbon observed in the effluent of these two media as compared to the other two is the most likely cause of the quick reduction in dissolved selenium concentrations. Over longer periods of time, Medium 5 may begin to have higher dissolved organic matter as the coconut fiber (coir) breaks down, resulting in greater selenium reduction. Medium 6 has the potential of being a viable selenium-reducing medium if more acetate were added. The mass of acetate that was added to Medium 6 was based on estimates of the amount

of bioavailable iron in the silica sand and other electron acceptors in the source water (i.e. DO, selenium, and manganese). The concentrations of manganese in the effluent relative to the concentrations in the influent indicate that a large mass of manganese may be present in the silica sand, which was not accounted for, that effectively "consumed" the small quantity of acetate that was added.

Recommendations For Further Research

Based on the analysis of the results of the column experiment it is recommended that Medium 3 be used for further study in the flow-through

Table 2. Percent increase in dissolved selenium concentrations relative to source water.				
Media	3	4	5	6
% Increase Relative to New Source Water	8.2%	32.1%	120.2%	32.6%

mesocosm experiment. The green waste mixture had comparable results, but it is not as favorable due to the variability that municipal "green waste" may have (i.e. could be grass clippings, woody debris, leaf litter, etc.). Also, "green waste" has the potential of containing undesirable chemicals such as pesticides and herbicides. Furthermore, Medium 3 had less of a response to the influx of fresh source water than Medium 4, indicating that it is a better candidate for the flow-through mesocosm experiment. It is believed that the addition of acetate, as was used in Medium 6, may be useful if, during the mesocosm experiment, dissolved organic carbon is observed to decrease.

Several lessons learned during the column experiment lead to the following recommendations:

- Perform hydraulic conductivity and effective porosity tests of the native soils to ensure that adequate flow rates and volume will be available for sampling.
- Measure the bioavailable iron and manganese in the native soils and sand so a more accurate estimate of the amount of electron donor needed to reduce the influent selenium concentrations can be calculated.
- Include a dissolved oxygen sampling port that gives a direct measurement within the soil matrix rather than in the effluent.
- Increase the frequency of sampling and/or number of samples to estimate rates of geochemical transformations or other trends.

References

Hibbs, Barry J. and M.M. Lee (2000). "Sources of Selenium in the San Diego Creek Watershed", *Final report to Defend the Bay and the California Urban Environmental Research & Education Center*, Department of Geological Sciences, CSULA.

Masscheleyn, P.H., R.D. Delaune, and W.H. Patrick, Jr. (1991). "Biogeochemical Behavior of Selenium in Anoxic Soils and Sediments: An Equilibrium Thermodynamics Approach." *Environ. Sci. Health*. A26(4): 555-573. **ERESA**

EDUCATING PROFESSIONALS IN CONSTRUCTED WETLAND TECHNOLOGY

-or-

"...ISN'T A SEPTIC LEACHFIELD JUST A VEGITATED GRAVELBED...?"

by

Daryl Van Dyke

Grad Student in Environmental Systems/IDT

This past May, the 2002 Constructed Wetlands Workshop drew a capacity crowd of professionals representing many fields, countries of origin, and stakeholder interests. Practicing engineers, wastewater treatment plant managers, wastewater treatment plant operators, bioremediation professionals, and representatives of public and private wildlife and conservation interests came together on the HSU campus to interact with Dr. Bob Gearheart and invited speakers.

The workshop was set for three days, and we were fortunate to have an unseasonable sunny and warm early Humboldt summer. On day one, after coffee, juice, and some of Barbara's delicious sweet-breads, Bob was on top for overviews (both hydrologic and biotic) which segued into design criteria. By the afternoon, various voices that shared interests or points of view cultivated a dialogue on the design and implementation of constructed wetlands to achieve a range of ecosystem and waste treatment benefits.

Bob fielded a wide range of questions about various facets of wetland and conventional wastewater treatment, responding with characteristic knowledge, candor and humor about ecological function at the center of wetland design. At one point, a discussion of field application for adsorption and infiltration as a disposal/recharge strategy veered toward septic leachfield design; Bob

noted the connection and steered us back to the issue at hand.

The program included talks by (ERE alums) Laura Kadlecik and Eric Strecker on stormwater treatment applications to constructed wetlands, and (ERE alum) Mike Wilson led a tour of the Potawatow Indian Health Village stormwater wetland system. Fellow ERE alum George Waller discussed the selection of plant species for wetland construction. Yociel Marrero Baez from Cuba gave an educational account of pursuing ecological reactors in a peri-urban Havana.

By day three, the atmosphere in the hall was friendly but focused. Folks had settled into their spots in the room and we were ready to talk wetlands. Bob Bastion of the EPA finished up with audience questions on the regulatory aspects of wetland implementation, pulling together a list of questions and concerns and distilling the reasons behind the regulations. Several field trips emphasized points that design criteria and plots of nitrogen distributions can't: wetlands as habitat, wetlands as waste treatment and recreational area simultaneously, and how our wetlands interface with municipal Arcata flows into the surrounding watershed.

The workshop concluded with a group photo. The crowd dispersed, visions of birds and suspended/attached microbial communities dancing in our heads. We won't look at a marsh or a wastewater treatment plant in the same way again. **ERESA**

ERE Curriculum Changes

• cont. inued from page 4

per week. In addition, the name of ERE elective course ENGR 421 has been changed, from "Computational Methods III" to "Computational Methods IV." The new courses cover essentially the same topics as the old courses, but the material has been substantially rearranged.

The primary purpose of the change is to separate linear algebra and ordinary differential equations so that each can be taught with the proper prerequisites at the right level of sophistication and at the right time in the sequence with other ERE major courses. It will also allow us to better integrate environmental resources engineering applications into the teaching of differential equations so students will be better prepared to use the material in later ERE courses.

Table 1 shows how the new ENGR 225, 325, 326 "Computational Methods" series is sequenced with other key ERE and Math courses in the ERE major curriculum. MATH 115 "Algebra and Elementary Functions" is not required, but is included because most ERE students take it to prepare for calculus.

Table 1. Sequencing of ENGR 225, 325, 326 with key ERE and Math courses in ERE major

Semester 1	Semester 2
MATH 115 ENGR 115 ENGR 225	MATH 109 ENGR 215
Semester 3	Semester 4
MATH 110 ENGR 210 ENGR 323 ENGR 325	MATH 210 ENGR 211
Semester 5	Semester 6
ENGR 313 ENGR 326 ENGR 331(C)	ENGR 333

The New ERE Curriculum

Figure 1 shows the new requirements for the ERE major, effective Spring 2003. Changes are in bold print. **ERESA**

ENVIRONMENTAL RESOURCES ENGINEERING

Requirements for the Major

A minimum grade of C- is required for all courses. Engineering courses may not be repeated more than two times.

LOWER DIVISION

MATH 109, 110, 210	Calculus I, II, III	4, 4, 4
CHEM 109, 110	General Chemistry I, II	5, 5
BIOL 105	Principles of Biology	4
PHYX 110	General Physics II	4
ENGR 115	Introduction to ERE	3
ENGR 210	Solid Mechanics: Statics	3
ENGR 211	Solid Mechanics: Dynamics	3
ENGR 215	Introduction to Design	3
ENGR 225	Computational Methods I	3

UPPER DIVISION

PHYX 315	Electronic Instrumentation	3
ENGR 313	Systems Analysis	4
ENGR 323	Probabilistic Analysis of Envir Systems	3
ENGR 324	Envir Monitoring & Data Analysis	3
ENGR 325	Computational Methods II	3
ENGR 326	Computational Methods III	3
ENGR 330	Mechanics & Science of Materials	3
ENGR 331	Thermodynamics & Energy Systems I	3
ENGR 333	Fluid Mechanics	4
ENGR 350	Introduction to Water Quality	3
ENGR 353	Environmental Health Engineering	3
ENGR 410	Environmental Impact Assessment	3
ENGR 416	Transport Phenomena	3
ENGR 435	Solid Waste Management	3
ENGR 440	Hydrology I	3
ENGR 492	Capstone Design Project	3

Major Elective Program

MAJOR ELECTIVE PROGRAM

With advice and approval of an ERE faculty advisor and the ERE Department Chair, select one upper division science or natural resources course and three senior engineering design courses from the following lists to form a coherent elective program.

One science/NR course from: (See catalog for list of courses.)

Three engineering design courses from:

ENGR 418	Applied Hydraulics	3
ENGR 421	Computational Methods III	3
ENGR 431	Air Quality Engineering	3
ENGR 441	Hydrology II	3
ENGR 443	Groundwater Systems	3
ENGR 445	Water Resources Planning & Mgt	3
ENGR 448	River Hydraulics	3
ENGR 451	Water & Waste Water Treatment Engr	3
ENGR 461	Environmental Geotechnology	3
ENGR 466	Earthquake Engineering	3
ENGR 471	Thermodynamics & Energy Systems II	3
ENGR 473	Building Energy Analysis	3
ENGR 475	Renewable Energy Power Systems	3
ENGR 477	Solar Thermal Engineering	3
ENGR 481	Selected Topics w/ Design	3
ENGR 498	Directed Design Project	3

FIGURE 1. New (Spring 2003) Requirements for the ERE Major (Changes in Bold)

Thanks for Making It Possible...



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Today . .

Tomorrow . .

Forever.



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